## **Biomolecules**

## FT Self Evaluation Test -31

- 1. Which does not show mutarotation
  - (a) Sucrose
- (b) Maltose
- (c) Glucose
- (d) Fructose
- 2. Artificial silk is
  - (a) Polyamides
- (b) Polyesters
- (c) Polyacids
- (d) Polysaccharides
- **3.** Which of the following is a protein
  - (a) Pepsin
- (b) Adrenaline
- (c) ATP
- (d) Glutamin
- 4. Glucose gives many reactions of aldehyde,

[CPMT 1977]

- (a) It is hydrolysed to acetaldehyde
- (b) It is a polyhydroxy ketone
- (c) It is a cyclic aldehyde
- (d) It is a hemiacetal in equilibrium with its aldehyde form in solution
- Glucose in blood can be quantitatively determined with

[JIPMER 2002]

- (a) Tollen's reagent
- (b) Benedict's solution
- (c) Alkaline iodine solution
- (d) Bromine water
- 6. Which of the following ions can cause coagulation of proteins [Kerala (Med.) 1999]
  - (a) *Na*<sup>+</sup>
- (b)  $Ag^+$
- (c) Ca ++
- (d)  $Mg^{++}$
- 7. Glucose reacts with methyl alcohol to give [CPMT 1985]
  - (a)  $\alpha$  methyl glucoside
  - (b)  $\beta$  methyl glucoside
  - (c) Both (a) and (b)
  - (d) None of these
- 8. Molisch's test is done for the detection of [BHU 1987]
  - (a) Alkyl halide
- (b) Carbohydrate
- (c) Alkaloid
- (d) Fat
- 9. Which of the following is not an amino acid

[MP PET/PMT 1998]

- (a) Glycine
- (b) Alanine
- (c) Histidine
- (d) Benzidine
- **10.** A substance forms zwitter ion. It can have functional groups

[DCE 2002]

- (a)  $-NH_2$ ,-COOH
- (b)  $-NH_2$ ,  $-SO_3H$

[bg:66EH07903]

- (d) None of these
- **11.** Which functional group participates in disulphide bond formation in proteins
  - (a) Thiolactone
- (b) Thiol
- (c) Thioether
- (d) Thioester
- **12.** Schweitzer's reagent used for dissolving cellulose in the manufacture of artificial silk is[Roorkee 1999]
  - (a)  $CuSO_4.5H_2O$
  - (b) CuI
  - (c)  $[Cu(NH_3)_4]SO_4$
  - (d)  $Cu(CH_3COO)_2$ .  $Cu(OH)_2$
- **13.** Which one of the following statements is true for protein synthesis (translation) [AIIMS 2005]
  - (a) Amino acid are directly recognized by m-RNA
  - (b) The third base of the codon is less specific
  - (c) Only one codon codes for an amino acid
  - (d) Every *t*-RNA molecule has more than one amino acid attachment site.
- **14.** In both DNA and RNA, heterocyclic base and phosphate ester linkages are at [AIEEE 2005]
  - (a)  $C_5'$  and  $C_2'$  respectively of the sugar molecule
  - (b)  $C_2'$  and  $C_5'$  respectively of the sugar molecule
  - (c)  $C'_1$  and  $C'_5$  respectively of the sugar molecule
  - (d)  $C_5'$  and  $C_1'$  respectively of the sugar molecule
- **15.** Which of the following biomolecules contain non-transition metal ion [KCET 2005]
  - (a) Vitamin  $B_{12}$
- (b) Chlorophyll
- (c) Haemoglobin
- (d) Insulin
- **16.** An example of a sulphur containing amino acid is [KCET 2005]
  - (a) Lysine
- (b) Serine
- (c) Cysteine
- (d) Tyrosine





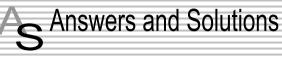


## **Biomolecules 1469**

- **17.** Which of the following is not present in a nucleotide
- (c) Adenine
- (d) Tyrosine

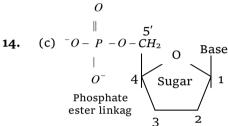
[KCET 2005]

- (a) Cytosine
- (b) Guanine



(SET -31)

- **1.** (a) Sucrose does not show mutarotation due to non reducing nature.
- **2.** (d) It is a polysaccharide.
- 3. (a) Pepsin is a protein.
- **4.** (d) It is a hemiacetal in equilibrium with its aldehyde form in solution.
- **5.** (a) In glucose aldehydic group is present and Tollen's reagent is the test for aldehydes.
- **6.** (b)  $Ag^+$  can cause coagulation of proteins.
- **7.** (c) Alpha methyl glucoside and beta methyl glucoside.
- **8.** (b) Molisch's test is done for the detection of carbohydrate bond formation.
- 9. (d) Benzidine is not an amino acid. It is an amine.
- 10. (c) A substance forms Zwitter ion. It can have functional groups  $-NH_2COOH$  and  $-NH_2$ ,  $-SO_3H$ .
- **11.** (b) Thiol functional group particpitates in disulphide in proteins.
- 12. (c)  $\left[Cu\left(NH_3\right)_4\right]SO_4$  is schweitzer's reagent used for manufacture of artificial silk.
- **13.** (a) In the process of translation amino acids are directly recognized by *m*-RNA.



**15.** (b) Biomolecule Metal ion

Vitamin  $B_{12}$ 

Co (transition metal)

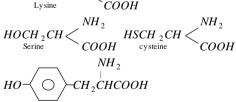
Chlorophyll

Mg (non-transition metal

ion)

Haemoglobin Insulin Fe (transition metal)
S (non-Metal)

**16.** (c)  $H_2N(CH_2)_4CH < NH_2$ 



\*\*\* 17

(d) Nucleotide contains nitrogenous bases like adenine, guanine, thymin, cytosine and uracil.



